

**Cubebin, a lignan isolated from *Aristolochia odoratissima* L.**

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**Key indicators**

Single-crystal X-ray study  
 $T = 291\text{ K}$   
Mean  $\sigma(\text{C}-\text{C}) = 0.006\text{ \AA}$   
 $R$  factor = 0.053  
 $wR$  factor = 0.121  
Data-to-parameter ratio = 9.4

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The structure of cubebin [systematic name: 3,4-bis(1,3-benzodioxol-5-ylmethyl)tetrahydrofuran-2-ol],  $\text{C}_{20}\text{H}_{20}\text{O}_6$ , is stabilized by  $\text{O}-\text{H}\cdots\text{O}$  and  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonds.

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**Comment**

Cubebin is a dibenzylbutyrolactone lignan, commonly found in coniferae and in the wood, roots and resin of many plants. It has been isolated from hinokiol (Ishiguro, 1936) and obtained by synthesis (Battersbee *et al.* 1969). Cubebin and its derivatives show a range of biological activities; recent studies report their analgesic, anti-inflammatory and trypanocidal activity (Bastos *et al.*, 2001; Souza *et al.*, 2004; de Souza *et al.*, 2005; da Silva *et al.*, 2005). Our material was isolated from *Aristolochia odoratissima* L. (Usobilaga *et al.*, 2005), which grows in the humid lowlands of the Maracaibo Lake (Venezuela). This plant contains aristolochic acid and is used as an anti-ophidian remedy.

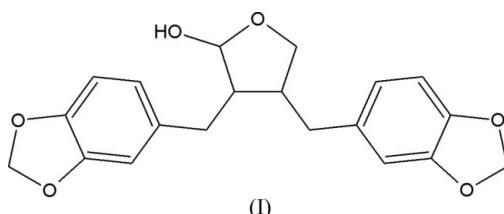


Fig. 1 shows the molecular structure of the title compound, (I), with the atom- and ring-labelling scheme. The five-membered ring C adopts an envelope conformation, with C2 as the flap atom. The substituents at C2 and C3 on ring C are *trans* to each other.

A search of the Cambridge Structural Database (Version 5.28; Allen, 2002) resulted in only one similar structure, a lignan with the same molecular formula isolated from *Daphne tangutica* Maxim, called (–)-dihydrosesamin, (II) (Lin-Gen *et al.*, 1983).

**Experimental**

Cubebin was extracted from the ground roots of *Aristolochia odoratissima* L. by treatment with *n*-pentane. Upon concentration of the pentane extract to half its volume, a precipitate was obtained which was filtered and crystallized from  $\text{CHCl}_3$ –hexane (1:1). Recrystallization from pentane yielded colourless needles (140 mg; m.p. 403–406 K). Mass spectrum:  $M^+$  356 ( $\text{C}_{20}\text{H}_{20}\text{O}_6$ , 32%), 338 ( $M^+ - \text{H}_2\text{O}$ , 12%), 203 (15%), 135 (100%).  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ,  $\delta$ , p.p.m.): 6.68 (*d*, 2H, H11 and H19), 6.62 (*m*, 2H, H12 and H20), 6.47 (*s*, 2H, H7 and H15), 5.88 (*s*, 4H,  $\text{OCH}_2\text{O} \times 2$ ), 5.12 (*d*, 1H, H1), 3.90 (*t*, 1H, H4A), 3.72 (*t*, 1H, H4B), 2.64 (*m*, 4H, H13A,B and H5A,B), 2.09 (*m*, 1H, H3) 2.06 (*m*, 1H, H2).

**Crystal data**

$C_{20}H_{20}O_6$   
 $M_r = 356.36$   
Monoclinic,  $P2_1$   
 $a = 11.631 (2) \text{ \AA}$   
 $b = 5.5969 (10) \text{ \AA}$   
 $c = 13.577 (2) \text{ \AA}$   
 $\beta = 99.548 (3)^\circ$

$V = 871.6 (3) \text{ \AA}^3$   
 $Z = 2$   
Mo  $K\alpha$  radiation  
 $\mu = 0.10 \text{ mm}^{-1}$   
 $T = 291 \text{ K}$   
 $0.3 \times 0.2 \times 0.1 \text{ mm}$

**Data collection**

Bruker SMART CCD area-detector diffractometer  
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)  
 $T_{\min} = 0.944$ ,  $T_{\max} = 0.989$

5459 measured reflections  
2212 independent reflections  
962 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.070$

**Refinement**

$R[F^2 > 2\sigma(F^2)] = 0.053$   
 $wR(F^2) = 0.121$   
 $S = 0.99$   
2212 reflections  
235 parameters

1 restraints  
H-atom parameters constrained  
 $\Delta\rho_{\max} = 0.15 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\min} = -0.16 \text{ e \AA}^{-3}$

**Table 1**  
Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
O2—H2A $\cdots$ O1 <sup>i</sup>	0.82	1.95	2.744 (4)	164
C9—H9A $\cdots$ O4 <sup>ii</sup>	0.97	2.59	3.182 (6)	120
C13—H13A $\cdots$ O2	0.97	2.51	2.838 (5)	100

Symmetry codes: (i)  $-x + 1, y + \frac{1}{2}, -z + 2$ ; (ii)  $-x + 2, y + \frac{1}{2}, -z + 1$ .

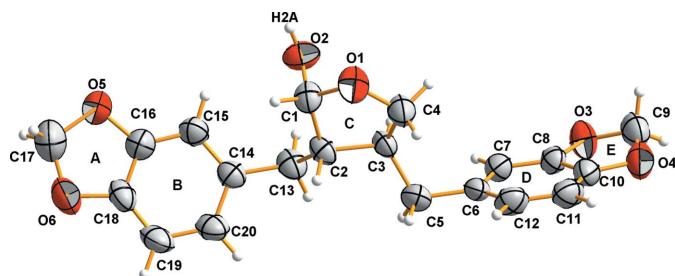
H atoms were placed in calculated positions and refined using a riding model, with  $C-H = 0.93-0.98 \text{ \AA}$  and  $O-H = 0.82 \text{ \AA}$ , and with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C}, \text{O})$ . In the absence of appreciable anomalous scattering, Friedel equivalents were merged before the final refinement cycles. The configuration reported in the literature was used for the refinement.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 1998); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *DIAMOND* (Brandenburg, 2001); software used to prepare material for publication: *PLATON* (Spek, 2003), *enCIFer* (Allen *et al.*, 2004) and *publCIF* (Westrip, 2007).

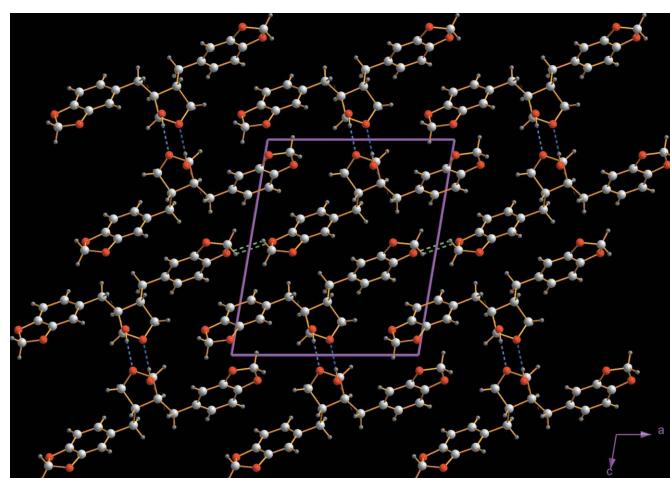
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**Figure 1**

The molecular structure of the title compound, with the atom- and ring-labelling scheme. Displacement ellipsoids are drawn at the 50% probability level.

**Figure 2**

View along the  $b$  axis of the packing arrangement and intermolecular hydrogen bonds for the title compound. Blue dashed lines indicate  $O2-H2A\cdots O1$  hydrogen bonds and green dashed lines  $C9-H9A\cdots O4$ .